3.1 CHEMICAL IDENTITY

Data pertaining to the chemical identity of 1,2-dibromo-3-chloropropane are listed in Table 3-1.

3.2 PHYSICAL AND CHEMICAL PROPERTIES

The physical and chemical properties of 1,2-dibromo-3-chloropropane are presented in Table 3-2.

TABLE 3-1. Chemical Identity of 1,2-Dibromo-3-chloropropane

Characteristic	Information	Reference
Chemical name	1,2-Dibromo-3-chloropropane	CAS 1989
Synonyms	DBCP; BBC12	CAS 1989
Trade names	Nemagon; Nemafume; Fumazone; Fumagon; Nemabrom; Nemazon; OS 1897; and others	OHM/TADS 1989
Chemical formula	C ₃ H ₅ Br ₂ Cl	CAS 1989
Chemical structure	CH ₂ CHCH ₂ I I \ Br Br Cl	CAS 1989
Identification numbers:		
CAS registry NIOSH RTECS EPA hazardous waste OHM/TADS DOT/UN/NA/IMCO shipping HSDB NCI	96-12-8 TX8750000 U066 7216513 UN 2872 1629 C00500	CAS 1989 HSDB 1989 HSDB 1989 OHM/TADS 1989 HSDB 1989 HSDB 1989

CAS = Chemical Abstracts Service; EPA = Environmental Protection Agency; DOT/UN/NA/IMCO = Department of Transportation/ United Nations/ North America/ International Maritime Dangerous Goods Code; HSDB = Hazardous Substances Data Bank; NCI = National Cancer Institute; NIOSH = National Institute for Occupational Safety and Health; OHM/TADS = Oil and Hazardous Materials/ Technical Assistance Data System; RTECS = Registry of Toxic Effects of Chemical Substances

TABLE 3-2. Physical and Chemical Properties of 1,2-Dibromo-3-chloropropane

Property	Information	Reference
Molecular weight	236.36	Windholz 1983
Color	Colorless (when pure); amber to dark brown, yellow (technical grade)	Sax and Lewis 1987; Verschueren 1983; NIOSH 1985
Physical state	Liquid	Windholz 1983
Melting point	6°C	Stenger 1978
Boiling point	196°C	Windholz 1983
Density at 14°C	2.093 g/cm ³	Windholz 1983
Odor	Pungent	Windholz 1983
Odor threshold:	3	
Water	No data	
Air	0.0965 mg/m ³	Ruth 1986
Solubility:	G ,	
Water at 20°C	1,230 mg/L	Munnecke and VanGundy 1979
Organic solvents	Miscible with methanol, ethanol, isopropyl alcohol, hydrocarbons, halogenated hydrocarbons, and oils	IARC 1979; Windholz 1983
Partition coefficients:		
Log K _{ow}	2.26 (estimated)	EPA 1988a
Log K _{oc}	2.17; 2.11	Sabljic 1984; Wilson et al. 1981
Bioconcentration factor	11.2ª	Bysshe 1982
Vapor pressure at 20°C	0.58 mmHg	Munnecke and VanGundy 1979
Henry's law constant:		
at 20°C	$1.47 \times 10^{-4} \text{ atm-m}^3/\text{mol}^b$	Thomas 1982
Autoignition temperature Flashpoint:	No data	
Open cup Flammability limits	76.6°C (170°F) No data	Sax and Lewis 1987

TABLE 3-2 (Continued)

Property	Information	Reference
Conversion factors:		
ppm (v/v) to mg/m^3 in air $(20^{\circ}C)$	1 ppm (v/v) x 9.67 = mg/m3	
mg/m^3 to ppm (v/v) in air $(20^{\circ}C)$	$1 \text{ mg/m}^3 \text{ x } 0.103 = \text{ppm } (\text{v/v})$	
Explosive limits	No data	

^aCalculated from water solubility using equation 5-3 in Lyman et al. 1982 ^bCalculated from vapor pressure and water solubility using equation 15-8 in Lyman et al. 1982